

# SDI Limited

Version No: 9.1.1.1 Safety Data Sheet (Conforms to Regulation (EC) No 2015/830) Issue Date: 24/05/2016 Print Date: 31/05/2016 Initial Date: Not Available L.REACH.GBR.EN

### SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

#### 1.1.Product Identifier

Product name	Pola Night 22% Carbamide Peroxide Gel	
Synonyms	Not Available	
Other means of identification	Not Available	

### 1.2. Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses	To remove discoloration of teeth, to be performed by a dentist.	
Uses advised against	Not Applicable	

### 1.3. Details of the supplier of the safety data sheet

Registered company name	SDI Limited	SDI Brazil Industria E Comercio Ltda	SDI Germany GmbH
Address	3-15 Brunsdon Street VIC Bayswater 3153 Australia	Rua Dr. Virgilio de Carvalho Pinto, 612 São Paulo CEP 05415-020 Brazil	Hansestrasse 85 Cologne D-51149 Germany
Telephone	+61 3 8727 7111 (Business Hours) +55 11 3092 7100 +49 0 2203 9255 0		+49 0 2203 9255 0
Fax	+61 3 8727 7222	+55 11 3092 7101	+49 0 2203 9255 200
Website	www.sdi.com.au	www.sdi.com.au	www.sdi.com.au
Email	info@sdi.com.au	brasil@sdi.com.au	germany@sdi.com.au
Registered company name	SDI (North America) Inc.		
Address	1279 Hamilton Parkway IL Itasca 60143 United States		
Telephone	+1 630 361 9200 (Business hours)		
Fax	Not Available		
Website	Not Available		
Email	USA.Canada@sdi.com.au		

### 1.4. Emergency telephone number

Association / Organisation	SDI Limited Not Available Not Available		Not Available
Emergency telephone numbers	+61 3 8727 7111	Not Available	Not Available
Other emergency telephone numbers	ray.cahill@sdi.com.au	Not Available	Not Available
Association / Organisation	Not Available		
Emergency telephone numbers	+61 3 8727 7111		
Other emergency telephone numbers	Not Available		

# SECTION 2 HAZARDS IDENTIFICATION

# 2.1.Classification of the substance or mixture

# Considered a hazardous mixture according to Reg. (EC) No 1272/2008 and their amendments. Not classified as Dangerous Goods for transport purposes.

DSD classification	In case of mixtures, classification has been prepared by following DPD (Directive 1999/45/EC) and CLP Regulation (EC) No 1272/2008 regulations	
DPD classification <sup>[1]</sup>	R36 Irritating to eyes.	
Legend:	1. Classification by vendor; 2. Classification drawn from EC Directive 67/548/EEC - Annex I ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI	

Classification according to regulation (EC) No 1272/2008 [CLP] <sup>[1]</sup>	Eye Irritation Category 2
Legend:	1. Classification by vendor; 2. Classification drawn from EC Directive 67/548/EEC - Annex I ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI
2.2. Label elements	
CLP label elements	
SIGNAL WORD	WARNING
Hazard statement(s)	
H319	Causes serious eye irritation.
Supplementary statement(s	5)
Not Applicable	
Precautionary statement(s)	Prevention
P280	Wear protective gloves/protective clothing/eye protection/face protection.
Precautionary statement(s)	Response
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P337+P313	If eye irritation persists: Get medical advice/attention.
Precautionary statement(s)	Storage
Not Applicable	
Precautionary statement(s)	) Disposal

### Not Applicable

# 2.3. Other hazards

Ingestion may produce health damage\*.

Cumulative effects may result following exposure\*.

REACh - Art.57-59: The mixture does not contain Substances of Very High Concern (SVHC) at the SDS print date.

### SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### 3.1.Substances

See 'Composition on ingredients' in Section 3.2

### 3.2.Mixtures

1.CAS No 2.EC No 3.Index No 4.REACH No	%[weight]	Name	Classification according to directive 67/548/EEC [DSD]	Classification according to regulation (EC) No 1272/2008 [CLP]
1.124-43-6 2.204-701-4 3.Not Available 4.Not Available	22	urea hydrogen peroxide	R8, R20/22, R34, R41 <sup>[1]</sup>	Oxidizing Solid Category 3, Metal Corrosion Category 1, Acute Toxicity (Oral) Category 4, Acute Toxicity (Inhalation) Category 4, Skin Corrosion/Irritation Category 1B, Serious Eye Damage Category 1; H272, H290, H302, H332, H314, H318 <sup>[1]</sup>
		equivalent to:		
1.7722-84-1 2.231-765-0 3.008-003-00-9 4.01-2119485845-22-XXXX	7.3	<u>hydrogen</u> peroxide	R5, R8, R20/22, R35 <sup>[2]</sup>	Oxidizing Liquid Category 1, Acute Toxicity (Inhalation) Category 4, Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 1A; H271, H332, H302, H314 <sup>[3]</sup>
Legend:		ion by vendor; 2. C ion drawn from C&i		ive 67/548/EEC - Annex I ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI

# SECTION 4 FIRST AID MEASURES

### 4.1. Description of first aid measures

General	If skin or hair contact occurs: Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation. If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel. If fumes, aerosols or combustion products are inhaled remove from contaminated area. Other measures are usually unnecessary.
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Continued...

# Pola Night 22% Carbamide Peroxide Gel

	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> </ul>
Eye Contact	<ul> <li>If this product comes in contact with the eyes:</li> <li>Wash out immediately with fresh running water.</li> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
Skin Contact	If skin or hair contact occurs: ► Flush skin and hair with running water (and soap if available). ► Seek medical attention in event of irritation.
Inhalation	<ul> <li>If fumes, aerosols or combustion products are inhaled remove from contaminated area.</li> <li>Other measures are usually unnecessary.</li> </ul>
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> </ul>

# $\ensuremath{\textbf{4.2}}$ Most important symptoms and effects, both acute and delayed

See Section 11

#### 4.3. Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

### SECTION 5 FIREFIGHTING MEASURES

### 5.1. Extinguishing media

- ▶ Foam.
- Dry chemical powder.
- BCF (where regulations permit).
- Carbon dioxide.
- Water spray or fog Large fires only.

#### 5.2. Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid any contamination of this material as it is very reactive and any contamination is potentially hazardous
5.3. Advice for firefighters	
Fire Fighting	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>Fight fire from a safe distance, with adequate cover.</li> <li>Extinguishers should be used only by trained personnel.</li> <li>Use water delivered as a fine spray to control fire and cool adjacent area.</li> <li>Avoid spraying water onto liquid pools.</li> <li>DO NOT approach containers suspected to be hot.</li> <li>Cool fire exposed containers with water spray from a protected location.</li> <li>If safe to do so, remove containers from path of fire.</li> <li>If fire gets out of control withdraw personnel and warn against entry.</li> <li>Equipment should be thoroughly decontaminated after use.</li> </ul>
Fire/Explosion Hazard	<ul> <li>Non combustible.</li> <li>Not considered a significant fire risk, however containers may burn.</li> <li>Decomposition may produce toxic fumes of; nitrogen oxides (NOx) carbon monoxide (CO) carbon dioxide (CO2)</li> </ul>

# SECTION 6 ACCIDENTAL RELEASE MEASURES

6.1. Personal precautions, protective equipment and emergency procedures

# See section 8

### 6.2. Environmental precautions

See section 12

### 6.3. Methods and material for containment and cleaning up

Minor Spills	Wipe with absorbent towel. Wash with water (15 mins).		
Major Spills	<ul> <li>Clear area of personnel and move upwind.</li> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> <li>Wear full body protective clothing with breathing apparatus.</li> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> <li>No smoking, flames or ignition sources. Increase ventilation.</li> <li>Contain spill with sand, earth or other clean, inert materials.</li> <li>NEVER USE organic absorbents such as sawdust, paper or cloth.</li> <li>Use spark-free and explosion-proof equipment.</li> <li>Collect any recoverable product into labelled containers for possible recycling.</li> </ul>		

- ► Avoid contamination with organic matter to prevent subsequent fire and explosion.
- DO NOT mix fresh with recovered material.
- Collect residues and seal in labelled drums for disposal.
- Wash area and prevent runoff into drains.
- Decontaminate equipment and launder all protective clothing before storage and re-use.
- If contamination of drains or waterways occurs advise emergency services.

#### 6.4. Reference to other sections

Personal Protective Equipment advice is contained in Section 8 of the SDS.

### SECTION 7 HANDLING AND STORAGE

### 7.1. Precautions for safe handling

Safe handling	<ul> <li>Avoid personal contact and inhalation of dust, mist or vapours.</li> <li>Provide adequate ventilation.</li> <li>Aways wear protective equipment and wash off any spillage from clothing.</li> <li>Keep material away from light, heat, flammables or combustibles.</li> <li>Keep cool, dry and away from incompatible materials.</li> <li>Avoid physical damage to containers.</li> <li>DO NOT repack or returm unused portions to original containers. Withdraw only sufficient amounts for immediate use.</li> <li>Use only minimum quantity required.</li> <li>Avoid physical damage in volatile solvents. Solvent evaporation should be controlled to avoid dangerous concentration of the peroxide.</li> <li>Do NOT allow peroxides to containers.</li> <li>Do NOT use meat spatulas to handle peroxides</li> <li>Do NOT use meat spatulas to handle peroxides</li> <li>Do NOT use glass containers with screw cap lids or glass stoppers.</li> <li>Store peroxides at the lowest possible temperature, consistent with their solubility and freezing point.</li> <li>CAUTION: Do NOT stel liquids or solutions of peroxides is at engreature below that at which the peroxide freezes or precipitates. Peroxides in this form are extremely shock and heat-sensitive. Refrigerated storage of peroxides must ONLY be in explosion-proof units.</li> <li>The hazards and consequences of fires and explosions during synthesis and use of peroxides is widely recognised; spontaneous or induced decomposition of an energy-rich compound causes a rise in the surrounding temperature; the temperature will rise until thermal balance is established or until the material heats there is consplicing reason to do otherwise, peroxide cannot be controlled and the area should be completed prior to heating and with good agitation.</li> <li>The most effective means for minimising the consequences of an accident is to limit quantities to a practical minimum. Even gram-scale explosions can be serious. Once jupited the burning of peroxides cannot be controlled and the area should be ecoxpleted</li></ul>
Fire and explosion protection	See section 5
Other information	Store between 2 and 25 deg C. <b>Do not</b> store in direct sunlight. Store in a dry and well ventilated-area, away from heat and sunlight.
2. Conditions for safe s	torage, including any incompatibilities
Suitable container	DO NOT repark. Use containers supplied by manufacturer only

Suitable container	DO NOT repack. Use containers supplied by manufacturer only.
Storage incompatibility	Avoid strong bases.

# 7.3. Specific end use(s)

See section 1.2

# SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

### 8.1. Control parameters

#### DERIVED NO EFFECT LEVEL (DNEL)

Not Available

### PREDICTED NO EFFECT LEVEL (PNEC)

### Not Available

### OCCUPATIONAL EXPOSURE LIMITS (OEL)

### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
UK Workplace Exposure Limits (WELs)	hydrogen peroxide	Hydrogen peroxide	1.4 mg/m3 / 1 ppm	2.8 mg/m3 / 2 ppm	Not Available	Not Available

#### EMERGENCY LIMITS

Ingredient	Material name	TEEL-1	TEEL-2	TEEL-3
urea hydrogen peroxide	Urea peroxide; (Urea hydrogen peroxide)	1.2 mg/m3	13 mg/m3	79 mg/m3
hydrogen peroxide	Hydrogen peroxide	Not Available	Not Available	Not Available

hydrogen peroxide	Hydrogen peroxide - 30%	33 pp	pm	170 ppm	330 ppm
Ingredient	Original IDLH		Revised IDLH		
urea hydrogen peroxide	Not Available		Not Available		
hydrogen peroxide	75 ppm		75 [Unch] ppm		

# MATERIAL DATA

# 8.2. Exposure controls

6.2. Exposure controls					
	Engineering controls are used to remove a hazard or place a barrier bet effective in protecting workers and will typically be independent of worke The basic types of engineering controls are: Process controls which involve changing the way a job activity or process Enclosure and/or isolation of emission source which keeps a selected ha "removes" air in the work environment. Ventilation can remove or dilute a the particular process and chemical or contaminant in use. Employers may need to use multiple types of controls to prevent employer General exhaust is adequate under normal operating conditions. Local e exists, wear approved respirator. Correct fit is essential to obtain adequa contaminants generated in the workplace possess varying "escape" vel to effectively remove the contaminant.	r interactions to provide this l s is done to reduce the risk. azard "physically" away from an air contaminant if designe e overexposure. exhaust ventilation may be re ate protection. Provide adequ	high level of protection. the worker and ventilation that stra d properly. The design of a ventilation equired in specific circumstances. If uate ventilation in warehouse or clo	tegically "adds" and on system must match risk of overexposure osed storage areas. Air	
	Type of Contaminant:			Air Speed:	
	solvent, vapours, degreasing etc., evaporating from tank (in still air).			0.25-0.5 m/s (50-100 f/min)	
8.2.1. Appropriate	aerosols, fumes from pouring operations, intermittent container filling, acid fumes, pickling (released at low velocity into zone of active generation)		s, welding, spray drift, plating	0.5-1 m/s (100-200 f/min.)	
engineering controls	direct spray, spray painting in shallow booths, drum filling, conveyer lo zone of rapid air motion)	ading, crusher dusts, gas di	scharge (active generation into	1-2.5 m/s (200-500 f/min.)	
	grinding, abrasive blasting, tumbling, high speed wheel generated dus air motion).	sts (released at high initial ve	elocity into zone of very high rapid	2.5-10 m/s (500-2000 f/min.)	
	Within each range the appropriate value depends on:				
	Lower end of the range		Upper end of the range		
	1: Room air currents minimal or favourable to capture		1: Disturbing room air currents		
	2: Contaminants of low toxicity or of nuisance value only.		2: Contaminants of high toxicity		
	3: Intermittent, low production.		3: High production, heavy use		
	4: Large hood or large air mass in motion		4: Small hood-local control only		
	Simple theory shows that air velocity falls rapidly with distance away from the opening of a simple extraction pipe. Velocity generally decreases with the square of distance from the extraction point (in simple cases). Therefore the air speed at the extraction point should be adjusted, accordingly, after reference to distance from the contaminating source. The air velocity at the extraction fan, for example, should be a minimum of 1-2 m/s (200-400 f/min) for extraction of solvents generated in a tank 2 meters distant from the extraction point. Other mechanical considerations, producing performance deficits within the extraction apparatus, make it essential that theoretical air velocities are multiplied by factors of 10 or more when extraction systems are installed or used.				
8.2.2. Personal protection					
Eye and face protection	<ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may lenses or restrictions on use, should be created for each workplace chemicals in use and an account of injury experience. Medical and i readily available. In the event of chemical exposure, begin eye irrigal at the first signs of eye redness or irritation - lens should be removec Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivala</li> </ul>	or task. This should include first-aid personnel should be tion immediately and remove d in a clean environment only	a review of lens absorption and ad trained in their removal and suitabl contact lens as soon as practicable	sorption for the class of le equipment should be e. Lens should be removed	
Skin protection	See Hand protection below				
Hands/feet protection	<ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber 72rub2</li> </ul>				
Body protection	See Other protection below				
Other protection	<ul> <li>Overalls.</li> <li>PVC Apron.</li> <li>PVC protective suit may be required if exposure severe.</li> <li>Eyewash unit.</li> <li>Ensure there is ready access to a safety shower.</li> </ul>				
Thermal hazards	Not Available				

### **Respiratory protection**

Type B Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required. Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

Required Minimum Protection Factor	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
up to 10 x ES	B-AUS	-	B-PAPR-AUS / Class 1
up to 50 x ES	-	B-AUS / Class 1	-
up to 100 x ES	-	B-2	B-PAPR-2 ^

^ - Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Armonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

### 8.2.3. Environmental exposure controls

See section 12

# SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES

### 9.1. Information on basic physical and chemical properties

Appearance Clear gel with spearmint odour, soluble in water.

Physical state	Gel	Relative density (Water = 1)	1.1
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	5.9-6.9	Decomposition temperature	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water (g/L)	Miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

9.2. Other information

Not Available

### SECTION 10 STABILITY AND REACTIVITY

10.1.Reactivity	See section 7.2
10.2.Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable under normal handling conditions.</li> <li>Prolonged exposure to heat.</li> <li>Hazardous polymerisation will not occur.</li> </ul>
10.3. Possibility of hazardous reactions	See section 7.2
10.4. Conditions to avoid	See section 7.2
10.5. Incompatible materials	See section 7.2
10.6. Hazardous decomposition products	See section 5.3

# SECTION 11 TOXICOLOGICAL INFORMATION

### 11.1. Information on toxicological effects

Inhaled	The material is not thought to produce either adverse health effects or irritation of the respiratory tract following inhalation (as classified by EC Directives using animal models). Nevertheless, adverse systemic effects have been produced following exposure of animals by at least one other route and good hygiene practice requires that exposure be kept to a minimum and that suitable control measures be used in an occupational setting.
Ingestion	Accidental ingestion of the material may be damaging to the health of the individual.
Skin Contact	The material is not thought to produce adverse health effects or skin irritation following contact (as classified by EC Directives using animal models). Nevertheless, good hygiene practice requires that exposure be kept to a minimum and that suitable gloves be used in an occupational setting.
Eye	Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.
Chronic	Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems.

Pola Night 22% Carbamide Peroxide Gel	TOXICITY	IRRITATION	
	Not Available	Not Available	
	ТОХІСІТҮ	IRRITATION	
urea hydrogen peroxide	Not Available	Not Available	
	тохісіту	IRRITATION	
	dermal (rat) LD50: 3000-5480 mg/kg <sup>[1]</sup>	Nil reported	
hydrogen peroxide	Inhalation (rat) LC50: 2 mg/L/4H <sup>[2]</sup>		
	Oral (rat) LD50: 75 mg/kg <sup>[1]</sup>		
Legend:	nd: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances		

No significant curve tooloogical data identified in finature search.           UREA WYDROCCH         No significant curve tooloogical data identified in finature search.           UREA WYDROCCH         No significant curve tooloogical data identified in finature search.           UREA WYDROCCH         No significant curve tooloogical data identified in finature search.           American always dipercision providers or providers or providers in providers in finature search.         No significant curve tooloogical data identified in finature search.           American always dipercision always or providers in providers in finature search.         No significant curve to instantiation of the late of mathematic search.           American always dipercision always or providers in providers in the second always patient in curve search in the second always and the second always of providers in the curve intermediation.         No significant curve tool always and the second always and the						
UREA HYDROGEN PERCOND         Insection analysis of point can be solved by a present of the present of present a simulation of the can be solved by a present of the can be solved		No significant acute toxicological data identified in literature search.				
HYDROGEN PERCOVID       Address for months or even years after exposure to the material casese. This may be due to a non-allergenic condition known as nearbie anisway dydunction synchrone (RADS) which can council following exposure to thigh installing compound. Key criteria for the diagnosis to hour a dia documented exposure to the instant A nearbible anishou, an integrant document, which and a documented exposure to the instant A nearbible anishou and instant on a nearbible anishou and instant documents and the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of RADS, RADS for externing fibering an integrant document with mean restance of the diagnosis of regrant document with mean restance of the diagnosis of regrant document with mean restance of the diagnosis of regrant documents and the diagnosis of the diagnosis		reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production.				
HYDROGEN PERCIVICE <ul> <li>A constraint in dividual, with adapt or set of person presence of modeling sequences of the disposition invalues to hours of a documented exposure to the intertex. A neweable affior pattern, on spirometry, with the section of and duration of exposure to the intertex. A neweable affior pattern, on spirometry, with the section of and duration of exposure to the intertex. A neweable affior pattern, on spirometry, with the section of and duration of exposure to the intertex of a documented exposure to the intertex. A neweable affior pattern disorder with new relation to and duration of exposure to the intertex of a documentation. The other hand, a is a documentation is an interquent disorder with new relation of and duration of exposure to the intertex of the other hand, a is docided that occurs as result of apposing being of the pattern disorder. The other hand, a is docided that occurs are set of apposing to the high concentration of and duration of exposure to the intertex intertex of the other hand, a is docided that occurs as result of apposing the high composition of high composition of the other hand, a is docided that occurs as result of apposing the high composition of the other hand, a is addited in the set of the other hand, a is addited in the intertex in the other hand is in set addited in the intertex in the other hand is in the and tistus as obtained in mount appoint high tragen provide is composed in the bave before abarption. When applied to tessue, solutions of hydrogen perceide to hydrogen perceide is not addited in the intertex in the intertex of a documentation in high composition is the intertex in the other high of the intertex in thistice of the interex intertex in the other high of the intertex in</li></ul>		No significant acute toxicological data identified in literature s	search.			
HYDROGEN PEROXIDE       Pharmacokinetics         HYDROGEN PEROXIDE       Phidopan peroxide is a normal product of metabolically in intex cells and skiney, suggesting its distribution to those sites. Hydrogen peroxide has been detected in the table. Hydrogen peroxide has been detected in the product of metabolically in intex cells and tissues. It is formed by reduction of oxygen either directly in a two-electron transfer reaction, often calayleed by flexoproteins, or by an initial one-dectors site to 02 followed by disruption: Busines and mynas, liver, and kichey, may be distribution sites. In ables and cales that ded after intravenous administration of hydrogen peroxide, inclusion, such adad on the results of locabily studes, the lungs were pale and emphysematous. Flowing intraperitorial eitherion of the decomposing hydrogen peroxide, inclusion and cruss. Statistical on the results of locabily studes, the lungs. Neer, and kichey may be distribution sites. In ables and cases that ded after intravenous administration of hydrogen peroxide, the lungs were pale and emphysematous. Flowing intraperitorials eitherio distribution peroxide in the instative metabolically in intext cells and throme threats and emphysematous. Flowing intraperitorials eitherio distribution states, exponsible to decomposing hydrogen peroxide, intravenous administration of hydrogen peroxide, intravenous administration of hydrogen peroxide, intravenous administration of hydrogen peroxide, inclusion periodice of the composing hydrogen peroxide, intravenous administration of hydrogen peroxide, inclusion periodice of the composing hydrogen peroxide, induced to the cells. In compariso the hydrogen peroxide induced to the composing hydrogen peroxide, induced to administration of hydrogen peroxide, induced to the cells and three cells and information. The Solid to case and Aspergillis chovalline, but not to Strepatriprose grisofiavus. It was not mutagenic to Drosophila melanog		reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. Industrial bronchitis, on the other hand, is a disorder that occurs as result of exposure due to high concentrations of irritating substance (often particulate in nature) and is completely reversible after exposure ceases. The disorder is characterised by dyspnea, cough and mucus production. For hydrogen peroxide:				
HYDROGEN PEROXIDE       Peroxide, target organs affected include the lungs, intestine, thymus, liver, and kidney, suggesting its distribution to those sites.         HYDROGEN PEROXIDE       Peroxide is decomposed in the bowel before absorption. When applied to itssue, solutions of hydrogen peroxide have poor perioritability.         Bittability       Distribution, hydrogen peroxide is decomposed in the bowel before absorption. When applied to itssue, solutions of hydrogen peroxide.         HYDROGEN PEROXIDE       Hydrogen peroxide is postduard metabolically in initize of lead disease. It is formed by reduction of oxygen either directly in a two-electron transfer reaction, often catalysed by flexoproteins, or by an initial one-electron step to Q2 followed by dismutation to hydrogen peroxide.         HYDROGEN PEROXIDE       Hydrogen peroxide has been detected in serum and in match two: based on the results of thydrogen peroxide. The periods motion and in the time is and typical period is present in normal human tissues (IARC 1985). When hydrogen peroxide in human break list of the data present in normal human tissues (IARC 1985). When hydrogen peroxide in human break list of the data and enclaration approximation, a provide contex with catalase, a encoping for data most transmit for the fragment in more treated orally with hydrogen peroxide. Screentein hydrogen peroxide in human break list of the data and in hick three and the base detected in human. The state and thyme, like is in the indecend in the data and encocarinomas have been observed in mice treated orally with hydrogen peroxide. Material and the fragment in the fragment is the transmitter and thyme periodis induced prev		Pharmacokinetics				
HYDROGEN PEROXIDE		peroxide, target organs affected include the lungs, intestine, thymus, liver, and kidney, suggesting its distribution to those sites.				
HYDROGEN PEROXIDE <ul> <li></li></ul>		<ul> <li>Absorption: Hydrogen peroxide is decomposed in the bow</li> </ul>	vel before absorption. When applied	to tissue, solutions of hydrogen peroxide have poor		
Hydrogen peroxide has been detected in serum and in intact liver, based on the results of toxicity studies, the lungs, intesine, thymus, liver, and kidney may be distibution sites. In relation sites and cast that ide difer intravenous administration of hydrogen peroxide, the lungs were pale and emphysematous. Following intraperitoneal injection of hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide, is present in normal human tissues (IARC 1985). When hydrogen peroxide induced to dudenal telesions including adenomas, carcinomas, and adenocarcinomas have been observed in mice treated orally with hydrogen peroxide. Marked strain differences in the incidence of tumors have been observed. Papilloma development has been observed in mice treated by demal application. Genotoxicity         Hydrogen peroxide induced DNA damage, sister chromatid exchanges and chromosomal aberrations in mammalian cells in vitro. Hydrogen peroxide induced DNA damage in bacteria (E. col), and was mutagenic to <i>Drosophils melanogaster</i> or to mammalian cells in vitro.         Hydrogen peroxide induced DNA damage, sister chromatid exchanges and chromosomal aberrations in mammalian cells in vitro.       Hydrogen peroxide induced DNA damage.         Develop		Distribution Hydrogen peroxide is produced metabolically	•			
Gastric and duodenal lesions including adenomas, carcinomas, and adenocarcinomas have been observed in mice treated orally with hydrogen peroxide.         Marked strain differences in the incidence of tumors have been observed. Papilioma development has been observed in mice treated by dermal application.         Genotoxicity         Hydrogen peroxide induced DNA damage, sister chromatid exchanges and chromosomal aberrations in mammalian cells <i>in vitro</i> . Hydrogen peroxide induced DNA damage in bacteria ( <i>E. coli</i> ), and was mutagenic to bacteria ( <i>Salmonella typhimunium</i> ) and the fungi, <i>Neurospora crassa and Aspergillis chevallieri</i> , but not to <i>Streptorynees griseoflavus</i> . It was not mutagenic to <i>Drosophila melanogaster</i> or to mammalian cells <i>in vitro</i> .         Developmental Toxicity       Malformations have been observed in chicken embryos treated with hydrogen peroxide, but experiments with mice and rats have been negative.         Female rats that received 0.45% hydrogen peroxide (equivalent to approximately 630 mg/kg/day)7 as the sole drinking fluid for five weeks produced normal litters when mated with untreated males.         Doses of 1.4 to 11 molegg hydrogen peroxide (purity 30%) dissolved in water were injected into the airspace of groups of 20-30 white leghorn chicken eggs on day 3 of incubation.         Embryos were examined on day 14. The incidence of embryonic deaths and malformations was dose-related and detected at doses of 2.8 mol/egg and above. The combined ED50 was 2.7 mol/egg.         Reproductive Toxicity       A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertilly.         The solutance is classifie	HYDROGEN PEROXIDE	<ul> <li>Hydrogen peroxide has been detected in serum and in int distribution sites. In rabbits and cats that died after intrave intraperitoneal injection of hydrogen peroxide in mice, pyk renal tubular epithelial tissue was observed following oral</li> <li>Metabolism Glutathione peroxidase, responsible for deco peroxide comes in contact with catalase, an enzyme found</li> </ul>	act liver. based on the results of toxic enous administration of hydrogen per knotic nuclei were induced in the inte I administration of hydrogen peroxid omposing hydrogen peroxide, is pre- d in blood and most tissues, it rapidly	city studies, the lungs, intestine, thymus, liver, and kidney may be roxide, the lungs were pale and emphysematous. Following istine and thymus (IARC 1985). Degeneration of hepatic and e to mice. sent in normal human tissues (IARC 1985). When hydrogen decomposes into oxygen and water.		
Marked strain differences in the incidence of tumors have been observed. Papilloma development has been observed in mice treated by dermal application.         Genotoxicity         Hydrogen peroxide induced DNA damage, sister chromatid exchanges and chromosomal aberrations in mammalian cells <i>in vitro</i> . Hydrogen peroxide induced DNA damage in bacteria ( <i>E. coli</i> ), and was mutagenic to bacteria ( <i>Salmonella typhimunium</i> ) and the fungi, <i>Neurospora crassa and Aspergillis chevallieri</i> , but not to <i>Streptomyces griseoflavus</i> . It was not mutagenic to <i>Drosophila melanogaster</i> or to mammalian cells <i>in vitro</i> .         Developmental Toxicity         Malformations have been observed in chicken embryos treated with hydrogen peroxide, but experiments with mice and rats have been negative.         Female rats that received 0.45% hydrogen peroxide (equivalent to approximately 630 mg/kg/day)7 as the sole drinking fluid for five weeks produced normal litters when mated with untreated males.         Doses of 1.4 to 11 mol/egg hydrogen peroxide (purity 30%) dissolved in water were injected into the airspace of groups of 20-30 white leghorn chicken eggs on day 3 of incubation.         Embryos were examined on day 14. The incidence of embryonic deaths and malformations was dose-related and detected at doses of 2.8 mol/egg and above. The combined EDSO was 2.7 mol/egg.         Reproductive Toxicity       A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertity.         The substance is classified by IARC as Group 3:       NOT classifiable as to its carcinogenicity to humans.         Evidence of carcinogenicity may be ina		· ·	s, and adenocarcinomas have been	observed in mice treated orally with hydrogen peroxide.		
Hydrogen peroxide induced DNA damage, sister chromatid exchanges and chromosomal aberrations in mammalian cells <i>in vitro</i> . Hydrogen peroxide induced DNA damage in bacteria ( <i>E. coli</i> ), and was mutagenic to bacteria ( <i>Salmonella typhinnuium</i> ) and the fungi, <i>Neurospora crassa and Aspergillis chevallieri</i> , but not to <i>Streptomyces griseoflavus</i> . It was not mutagenic to <i>Drosophila melanogaster</i> or to mammalian cells <i>in vitro</i> .         Developmental Toxicity       Malformations have been observed in chicken embryos treated with hydrogen peroxide, but experiments with mice and rats have been negative.         Female rats that received 0.45% hydrogen peroxide (equivalent to approximately 630 mg/kg/day)? as the sole drinking fluid for five weeks produced normal litters when mated males.         Doses of 1.4 to 11 mol/egg hydrogen peroxide (equivalent to approximately 630 mg/kg/day)? as the sole drinking fluid for five weeks produced normal litters when mated ED50 was 2.7 mol/egg.         Reproductive Toxicity       A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility.         The substance is classified by IARC as Group 3:       NOT classifiable as to its carcinogenicity to humans.         Evidence of carcinogenicity may be inadequate or limited in animal testing.       Streptoductive Toxicity         Skin Irritation/Corrosion       Carcinogenicity       Store Streptoductive Store		Marked strain differences in the incidence of tumors have been				
Malformations have been observed in chicken embryos treated with hydrogen peroxide, but experiments with mice and rats have been negative.         Female rats that received 0.45% hydrogen peroxide (equivalent to approximately 630 mg/kg/day)7 as the sole drinking fluid for five weeks produced normal litters when mated with untreated males.         Doses of 1.4 to 11 mol/egg hydrogen peroxide (purity 30%) dissolved in water were injected into the airspace of groups of 20-30 white leghorn chicken eggs on day 3 of incubation.         Embryos were examined on day 14. The incidence of embryonic deaths and malformations was dose-related and detected at doses of 2.8 mol/egg and above. The combined ED50 was 2.7 mol/egg.         Reproductive Toxicity         A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility.         The substance is classified by IARC as Group 3:         NOT classifiable as to its carcinogenicity to humans.         Evidence of carcinogenicity may be inadequate or limited in animal testing.         Skin Irritation/Corrosion         Serious Eye         Serious Eye		Hydrogen peroxide induced DNA damage, sister chromatid e DNA damage in bacteria ( <i>E. coli</i> ), and was mutagenic to bac	teria (Salmonella typhimurium) and th	he fungi, Neurospora crassa and Aspergillis chevallieri, but		
Doses of 1.4 to 11 mol/egg hydrogen peroxide (purity 30%) dissolved in water were injected into the airspace of groups of 20-30 white leghorn chicken eggs on day 3 of incubation.         Embryos were examined on day 14. The incidence of embryonic deaths and malformations was dose-related and detected at doses of 2.8 mol/egg and above. The combined ED50 was 2.7 mol/egg.         Reproductive Toxicity         A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility.         The substance is classified by IARC as Group 3:         NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing.         Viete Toxicity         Skin Irritation/Corrosion         Serious Eye         Serious Eye		Malformations have been observed in chicken embryos treated Female rats that received 0.45% hydrogen peroxide (equivaler		•		
The combined ED50 was 2.7 mol/egg.         Reproductive Toxicity         A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility.         The substance is classified by IARC as Group 3:         NOT classifiable as to its carcinogenicity to humans.         Evidence of carcinogenicity may be inadequate or limited in animal testing.         Kin Irritation/Corrosion         Serious Eye		Doses of 1.4 to 11 mol/egg hydrogen peroxide (purity 30%) di	issolved in water were injected into t	he airspace of groups of 20-30 white leghorn chicken eggs on		
A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility.         The substance is classified by IARC as Group 3:         NOT classifiable as to its carcinogenicity to humans.         Evidence of carcinogenicity may be inadequate or limited in animal testing.         Acute Toxicity       O         Skin Irritation/Corrosion       Reproductivity         Serious Eye       STOT - Single Exposure	Embryos were examined on day 14. The incidence of embryonic deaths and malformations was dose-related and detected at doses of 2.8 mol/eg The combined ED50 was 2.7 mol/egg.					
NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing.         Acute Toxicity       Image: Carcinogenicity of the testing of testing o		A 1% solution of hydrogen peroxide (equivalent to 1900 mg/kg/day) given as the sole drinking fluid to three-month-old male mice for 7-28 days did not cause infertility. The substance is classified by IARC as Group 3: NOT classifiable as to its carcinogenicity to humans.				
Skin Irritation/Corrosion     Image: Skin Serious Eye       Serious Eye     STOT - Single Exposure						
Skin Irritation/Corrosion     Image: Skin Serious Eye       Serious Eye     STOT - Single Exposure	Acute Toxicitv	0	Carcinogenicity	0		
STOL-SINGLE EXPOSURE LIN	·					
	-	*	STOT - Single Exposure	0		

Respiratory or Skin sensitisation	$\odot$	STOT - Repeated Exposure	$\otimes$
Mutagenicity	0	Aspiration Hazard	0
		Ŭ 🗸	<ul> <li>Data available but does not fill the criteria for classification</li> <li>Data required to make classification available</li> <li>Data Not Available to make classification</li> </ul>

### SECTION 12 ECOLOGICAL INFORMATION

# 12.1. Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
hydrogen peroxide	LC50	96	Fish	0.020mg/L	3
hydrogen peroxide	EC50	3	Algae or other aquatic plants	0.27mg/L	4
hydrogen peroxide	EC50	48	Crustacea	2.32mg/L	4
hydrogen peroxide	EC50	72	Algae or other aquatic plants	0.71mg/L	4
hydrogen peroxide	NOEC	192	Fish	0.028mg/L	4
Legend:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

DO NOT discharge into sewer or waterways.

# 12.2. Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
hydrogen peroxide	LOW	LOW

### 12.3. Bioaccumulative potential

Ingredient	Bioaccumulation
hydrogen peroxide	LOW (LogKOW = -1.571)

### 12.4. Mobility in soil

Ingredient	Mobility
hydrogen peroxide	LOW (KOC = 14.3)

### 12.5.Results of PBT and vPvB assessment

	Р	В	т
Relevant available data	Not Available	Not Available	Not Available
PBT Criteria fulfilled?	Not Available	Not Available	Not Available

#### 12.6. Other adverse effects

No data available

### SECTION 13 DISPOSAL CONSIDERATIONS

#### 13.1. Waste treatment methods

Product / Packaging disposal	Consult manufacturer for recycling options. Consult State Land Waste Management Authority for disposal. Bury residue in an authorised landfill. Decontaminate empty containers.
Waste treatment options	Not Available
Sewage disposal options	Not Available

# SECTION 14 TRANSPORT INFORMATION

### Labels Required

Marine Pollutant	NO
HAZCHEM	Not Applicable
Land transport (ADR): NOT	T REGULATED FOR TRANSPORT OF DANGEROUS GOODS
14.1.UN number	Not Applicable
14.2.Packing group	Not Applicable
14.3.UN proper shipping name	Not Applicable
14.4.Environmental hazard	Not Applicable

14.5. Transport hazard class(es)	Class Not Applicable Subrisk Not Applicable	
	Hazard identification (Kemler)	Not Applicable
14.6. Special precautions for user	Classification code	Not Applicable
	Hazard Label	Not Applicable
	Special provisions	Not Applicable
	Limited quantity	Not Applicable

# Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.1. UN number	Not Applicable		
14.2. Packing group	Not Applicable		
14.3. UN proper shipping name	Not Applicable		
14.4. Environmental hazard	Not Applicable		
14.5. Transport hazard class(es)	ICAO/IATA ClassNot ApplicableICAO / IATA SubriskNot ApplicableERG CodeNot Applicable		
	Special provisions	Not Applicable	
	Cargo Only Packing Instructions	Not Applicable	
	Cargo Only Maximum Qty / Pack	Not Applicable	
14.6. Special precautions for user	Passenger and Cargo Packing Instructions	Not Applicable	
	Passenger and Cargo Maximum Qty / Pack	Not Applicable	
	Passenger and Cargo Limited Quantity Packing Instructions	Not Applicable	
	Passenger and Cargo Limited Maximum Qty / Pack	Not Applicable	

### Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.1. UN number	Not Applicable
14.2. Packing group	Not Applicable
14.3. UN proper shipping name	Not Applicable
14.4. Environmental hazard	Not Applicable
14.5. Transport hazard class(es)	IMDG Class     Not Applicable       IMDG Subrisk     Not Applicable
14.6. Special precautions for user	EMS Number     Not Applicable       Special provisions     Not Applicable       Limited Quantities     Not Applicable

# Inland waterways transport (ADN): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

14.1. UN number	Not Applicable		
14.2. Packing group	Not Applicable		
14.3. UN proper shipping name	Not Applicable		
14.4. Environmental hazard	Not Applicable		
14.5. Transport hazard class(es)	Not Applicable Not Applicable		
	Classification code Not Applicable		
	Special provisions Not Applicable		
· · ·	Limited quantity Not Applicable		
	Equipment required Not Applicable		
	Fire cones number Not Applicable		
14.6. Special precautions for user	Equipment required Not Applicable		

### Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

If packed as Chemical kits the following classification may be considered if all ICAO/IATA transport requirements are met: Chemical Kit UN3316 - Class 9.

### 15.1. Safety, health and environmental regulations / legislation specific for the substance or mixture

### UREA HYDROGEN PEROXIDE(124-43-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

European Customs Inventory of Chemical Substances ECICS (English)	European Union - European Inventory of Existing Commercial Chemical Substances (EINECS)
	(English)
HYDROGEN PEROXIDE(7722-84-1) IS FOUND ON THE FOLLOWING REGULATORY LIST	S
EU REACH Regulation (EC) No 1907/2006 - Annex XVII - Restrictions on the manufacture, placing on the market and use of certain dangerous substances, mixtures and articles	European Union (EU) Regulation (EC) No 1272/2008 on Classification, Labelling and Packaging of Substances and Mixtures - Annex VI
European Customs Inventory of Chemical Substances ECICS (English)	International Agency for Research on Cancer (IARC) - Agents Classified by the IARC
European Union - European Inventory of Existing Commercial Chemical Substances (EINECS)	Monographs
(English)	International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List

European Union (EU) Annex I to Directive 67/548/EEC on Classification and Labelling of Dangerous Substances - updated by ATP: 31

Passenger and Cargo Aircraft UK Workplace Exposure Limits (WELs)

This safety data sheet is in compliance with the following EU legislation and its adaptations - as far as applicable -: 67/548/EEC, 1999/45/EC, 98/24/EC, 92/85/EC, 94/33/EC, 91/689/EEC, 1999/13/EC, Commission Regulation (EU) 2015/830, Regulation (EC) No 1272/2008 and their amendments as well as the following British legislation: - The Control of Substances Hazardous to Health Regulations (COSHH) 2002 - COSHH Essentials - The Management of Health and Safety at Work Regulations 1999

#### 15.2. Chemical safety assessment

For further information please look at the Chemical Safety Assessment and Exposure Scenarios prepared by your Supply Chain if available.

#### ECHA SUMMARY

Ingredient	CAS number	Index No		ECHA Dossier	
urea hydrogen peroxide	124-43-6	Not Available		Not Available	
Harmonisation (C&L Inventory)	Hazard Class and Category Code(s)		Pictograms Signal Word Code(s)		Hazard Statement Code(s)
1	Ox. Sol. 3, Skin Corr. 1B		GHS05, GHS03, Dgr		H272, H314
2	Ox. Sol. 3, Skin Corr. 1B, Acute Tox. 4, Skin Irrit. 2, Eye Dam. 1, STOT SE 3, Ox. Sol. 2		GHS05, GHS03, Dgr		H272, H314, H302, H318, H335

Harmonisation Code 1 = The most prevalent classification. Harmonisation Code 2 = The most severe classification.

Ingredient	CAS number	Index No	ECHA Dossier			
hydrogen peroxide	7722-84-1 008-003-00-9 01-2		01-2	-2119485845-22-XXXX		
Harmonisation (C&L Inventory)	Hazard Class and Category Code(s)		Pictograms Signal Word Code(s)	Hazard Statement Code(s)		
1	Ox. Liq. 1, Acute Tox. 4, Skin Corr. 1A		GHS07, GHS05, GHS03, Dgr	H271, H302, H314, H332		
2	Ox. Liq. 1, Acute Tox. 4, Skin Corr. 1A, Eye Dam. 1, STOT SE 3, Aquatic Chronic 3, Ox. Liq. 2, Acute Tox. 3, Flam. Liq. 2, Skin Corr. 1B, Acute Tox. 2, Met. Corr. 1, Aquatic Chronic 2, Not Classified, Skin Irrit. 2, Eye Irrit. 2		GHS05, GHS03, Dgr, GHS02, GHS06, GHS09, Wng	H271, H314, H335, H318, H225, H301, H330, H290		

Harmonisation Code 1 = The most prevalent classification, Harmonisation Code 2 = The most severe classification,

National Inventory	Status
Australia - AICS	Y
Canada - DSL	N (urea hydrogen peroxide)
Canada - NDSL	N (hydrogen peroxide)
China - IECSC	Υ
Europe - EINEC / ELINCS / NLP	Υ
Japan - ENCS	N (urea hydrogen peroxide)
Korea - KECI	Y
New Zealand - NZIoC	Y
Philippines - PICCS	Υ
USA - TSCA	Υ
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

# **SECTION 16 OTHER INFORMATION**

#### Full text Risk and Hazard codes

H225	Highly flammable liquid and vapour.
H271	May cause fire or explosion; strong oxidiser.
H272	May intensify fire; oxidiser.
H290	May be corrosive to metals.
H301	Toxic if swallowed.
H302	Harmful if swallowed.
H314	Causes severe skin burns and eye damage.

H318	Causes serious eye damage.
H330	Fatal if inhaled.
H332	Harmful if inhaled.
H335	May cause respiratory irritation.
R20/22	Harmful by inhalation and if swallowed.
R34	Causes burns.
R35	Causes severe burns.
R41	Risk of serious damage to eyes.
R5	Heating may cause an explosion.
R8	Contact with combustible material may cause fire.

#### Other information

### DSD / DPD label elements



Relevant risk statements are found in section 2.1

Indication(s) of danger	Xi
SAFETY ADVICE	
S02	Keep out of reach of children.
\$23	Do not breathe gas/fumes/vapour/spray.
S26	In case of contact with eyes, rinse with plenty of water and contact Doctor or Poisons Information Centre.
S35	This material and its container must be disposed of in a safe way.
S39	Wear eye/face protection.
S40	To clean the floor and all objects contaminated by this material, use water.
S46	If swallowed, seek medical advice immediately and show this container or label.
S56	Dispose of this material and its container at hazardous or special waste collection point.
S64	If swallowed, rinse mouth with water (only if the person is conscious).

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by SDI Limited using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

For detailed advice on Personal Protective Equipment, refer to the following EU CEN Standards:

EN 166 Personal eye-protection

EN 340 Protective clothing

EN 374 Protective gloves against chemicals and micro-organisms

EN 13832 Footwear protecting against chemicals

EN 133 Respiratory protective devices

#### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

- PC-STEL: Permissible Concentration-Short Term Exposure Limit
- IARC: International Agency for Research on Cancer
- ACGIH: American Conference of Governmental Industrial Hygienists
- STEL: Short Term Exposure Limit
- TEEL: Temporary Emergency Exposure Limit。
- IDLH: Immediately Dangerous to Life or Health Concentrations
- OSF: Odour Safety Factor
- NOAEL :No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level
- LUMERI UNERI UNERIVEU AUVERSE ETTECT LE
- TLV: Threshold Limit Value LOD: Limit Of Detection

- OTV: Odour Threshold Value
- BCF: BioConcentration Factors
- BEI: Biological Exposure Index

The information contained in the Safety Data Sheet is based on data considered to be accurate, however, no warranty is expressed or implied regarding the accuracy of the data or the results to be obtained from the use thereof.

#### Other information:

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